Important Information about Your Drinking Water

The University of California, Davis tests the campus drinking water supply for many constituents as required by State and Federal Regulations. This report summarizes the results of our monitoring for the period of January 1 - December 31, 2019. This information is provided to inform the campus community about the monitoring and quality of the drinking water supply.

Este informe contiene información muy importante sobre su agua potable. Tradúzcalo o hable con alguien que lo entienda bien.

由於此報告書正包含着有關飲用水的重要信息。因此希望各位能夠翻譯或理解報告書內容的人對話。

Where Does Our Water Come From?

There are two drinking water sources used as the University’s water supply- Groundwater and Surface Water.

Groundwater

Six on-campus wells are used as the University’s groundwater supply. These wells draw water from aquifers 800 to 1,400 feet below the ground. The water is not treated, except for disinfection using chlorine (sodium hypochlorite). Chlorine levels are typically maintained between 0.50 and 1.0 ppm (parts per million) in the distribution system.

Surface Water

Starting July 2017, the University began to integrate treated surface water from the Sacramento River to the drinking water system. Raw surface water from the Sacramento River is taken in and pumped to the Regional Water Treatment Facility in Woodland. The raw water is treated by traditional surface water techniques, such as flash mixing and granular media filtration to remove microorganisms and other contaminants. The finished water is injected with chlorine and ortho-phosphate before it is delivered into the University's transmission line.

Substances That Might Be in Drinking Water

State-wide, the sources of drinking water (both tap water and bottled water) include rivers, lakes, streams, ponds, reservoirs, springs, and wells. As water travels over the surface of the land and through the ground, it dissolves naturally occurring minerals and, in some cases, radioactive material, and can pick up substances resulting from the presence of animals or from human activity.

In order to ensure that tap water is safe to drink, The US Environmental Protection Agency (USEPA) and the California State Water Resources Control Board, Division of Drinking Water prescribe regulations that limit the amount of certain contaminants in water provided by public water systems. Department regulations also establish limits for contaminants in bottled water, which must provide the same protection for public health. Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some substances. The presence of contaminants does not necessarily indicate that water poses a health risk.

Contaminants that may be present in source water include:

- **Microbial contaminants**, such as viruses and bacteria that may come from sewage treatment plants, septic systems, agricultural livestock operations, and wildlife.
- **Inorganic contaminants**, such as salts and metals, that can be naturally-occurring or result from urban stormwater runoff, industrial or domestic wastewater discharges, oil and gas production, mining, or farming.
- **Pesticides and herbicides** that may come from a variety of sources such as agriculture, urban stormwater runoff, and residential uses.
- **Organic chemical contaminants**, including synthetic and volatile organic chemicals, that are byproducts of industrial processes and petroleum production, and can also come from gas stations, urban stormwater runoff, agricultural application, and septic systems.
- **Radioactive contaminants** that can be naturally-occurring or be the result of oil and gas production and mining activities.

Drinking water, including bottled water, may reasonably be expected to contain at least small amounts of some contaminants. The presence of contaminants does not necessarily indicate that water poses a health risk. More information about contaminants and potential health effects can be obtained by calling the U.S. EPA’s Safe Drinking Water Hotline (1-800-426-4791).
Drinking Water Source Assessment Information

The drinking water assessments are intended to facilitate and provide the basic information necessary for a local community to develop a program to protect the drinking water supply.

Groundwater

A source assessment for the University’s groundwater wells was conducted in 2019. The University’s groundwater sources are considered most vulnerable to the following activities and not associated with any detected contaminants: Animal Activities; Sewer/Septic Collection & Treatment Systems; Pesticide/Fertilizer/Petroleum/Chemical Storage & Transfer Areas; and Research Laboratories. There have been no contaminants detected in the water supply attributable to these activities; however, the source is still considered vulnerable to activities located near the drinking water sources.

For more information, contact the University of California, Davis – Utilities Division. Contact information is provided below. Additional information can be found on the State Water Resources Control Board, Division of Drinking Water website at: http://www.waterboards.ca.gov/drinking_water/certlic/drinkingwater/DWSAP.shtml.

Surface Water

The surface water assessment for the Sacramento River watershed was conducted by several agencies. The Sacramento River Watershed Sanitary Survey 2015 Update Report can be found at: https://www.wdcwa.com/project-history/. The update stated that “Overall, the Sacramento River continued to provide good quality raw water. The raw water can currently be treated to meet all drinking water standards using conventional water treatment processes…” The report also identified eight key source water/watershed contaminant sources: Agricultural Drainage; Livestock; River Corridor and River Recreation; Homeless/Illegal Camping; Urban Runoff; Industrial National Pollutant Discharge Elimination System Permit Discharges; Wastewater Facilities; and Watershed Spills.

What Does Our Water Contain?

Tables 1 through 6 list the drinking water contaminants that were detected during the most recent sampling. The presence of these contaminants in the water does not necessarily indicate that the water poses a health risk. The Division of Drinking Water requires water suppliers to monitor for certain contaminants less than once per year because the concentrations of these contaminants are not expected to vary significantly from year to year. Some of the data, though representative of the water quality, is more than one year old.

In addition to the constituents listed in the tables below, our water was analyzed for numerous other substances that were below regulatory levels or not detectable. Additional information is available at the UC Davis Utilities website below and upon request: https://facilities.ucdavis.edu/utilities-water-quality-reports.

For more information, contact

<table>
<thead>
<tr>
<th>Name</th>
<th>Title</th>
<th>Phone</th>
<th>Email</th>
</tr>
</thead>
<tbody>
<tr>
<td>Joel McCoy</td>
<td>Superintendent - Water &amp; Gas</td>
<td>(530) 752-4825</td>
<td><a href="mailto:jymccoy@ucdavis.edu">jymccoy@ucdavis.edu</a></td>
</tr>
<tr>
<td>Michael Fan</td>
<td>Director - Utilities</td>
<td>(530) 752-7553</td>
<td><a href="mailto:mmfan@ucdavis.edu">mmfan@ucdavis.edu</a></td>
</tr>
</tbody>
</table>
### IMPORTANT DEFINITIONS:

The following are definitions of key terms for the water quality goals and standards noted on the data tables.

**Maximum Contaminant Level (MCL):** The highest level of a contaminant that is allowed in drinking water. Primary MCLs are set as close to the PHGs (or MCLGs) as is economically and technologically feasible. Secondary MCLs are set to protect the odor, taste, and appearance of drinking water.

**Maximum Contaminant Level Goal (MCLG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. MCLGs are set by the U.S. Environmental Protection Agency (USEPA).

**Public Health Goal (PHG):** The level of a contaminant in drinking water below which there is no known or expected risk to health. PHGs are set by the California Environmental Protection Agency.

**Maximum Residual Disinfectant Level (MRDL):** The highest level of a disinfectant allowed in drinking water. There is convincing evidence that addition of a disinfectant is necessary for control of microbial contaminants.

**Maximum Residual Disinfectant Level Goal (MRDLG):** The level of a drinking water disinfectant below which there is no known or expected risk to health. MRDLGs do not reflect the benefits of the use of disinfectants to control microbial contaminants.

**Primary Drinking Water Standards (PDWS):** MCLs or MRDLs for contaminants that affect health along with their monitoring and reporting requirements and water treatment requirements.

**Secondary Drinking Water Standards (SDWS):** MCLs for contaminants that affect taste, odor, or appearance of the drinking water. Contaminants with SDWSs do not affect health at the MCL levels.

**TOC:** Total Organic Carbon

**Regulatory Action Level (AL):** The concentration of a contaminant which, if exceeded, triggers treatment or other requirements that a water system must follow.

**Variance and Exemptions:** Department permission to exceed an MCL or not comply with a treatment technique under certain conditions.

**ND:** not detectable at testing limit

**ppm:** parts per million or milligrams per liter

**ppb:** parts per billion or micrograms per liter

**pCi/L:** picocuries per liter (a measure of radiation)

**μS/cm:** microsiemens per centimeter (a unit expressing the amount of electrical conductivity of a solution)

**UCD:** University of California, Davis

**USEPA:** The U.S. Environmental Protection Agency

### FOOTNOTES: MAJOR SOURCES IN DRINKING WATER

The following are footnotes referenced in the Major Sources in Drinking Water section of the data tables.

A. Erosion of natural deposits; residue from some surface water treatment processes
B. Erosion of natural deposits; runoff from orchards, glass and electronics production wastes
C. Discharges of oil drilling wastes and from metal
D. Runoff/leaching from natural deposits; seawater influence
E. Discharge from steel and pulp mills and chrome plating; erosion of natural deposits
F. Disinfection Byproducts
G. Internal corrosion of household plumbing systems; erosion of natural deposits; leaching from wood preservatives
H. Erosion of natural deposits; water additive that promotes strong teeth; discharge from fertilizer and aluminum factories
I. Sum of polyvalent cations present in the water, generally magnesium and calcium, and are usually naturally occurring
J. Leaching from natural deposits; industrial wastes
K. Internal corrosion of household water plumbing systems; discharges from industrial manufacturers; erosion of natural deposits
L. Leaching from natural deposits
M. Runoff and leaching from fertilizer use; leaching from septic tanks and sewage; erosion of natural deposits
N. Naturally-occurring organic materials
O. Herbicide runoff
P. Erosion of natural deposits
Q. Discharge from petroleum, glass, and metal refineries; erosion of natural deposits; discharge from mines and chemical manufacturers; runoff from livestock lots (feed additive)
R. Salt present in the water and is generally naturally occurring
S. Substances that form ions when in water; seawater influence
T. Runoff/leaching from natural deposits; industrial wastes
U. Coliforms are bacteria that are naturally present in the environment and are used as an indicator that other, potentially-harmful, bacteria may be present. Coliforms were found in more samples than allowed and this was a warning of potential problems.
V. Runoff/leaching from natural deposits
W. Soil runoff
### Table 1 - Detected Constituents With A Drinking Water Standard

<table>
<thead>
<tr>
<th>Constituent</th>
<th>Year Sampled</th>
<th>Unit</th>
<th>MCL (AL) [MRDL]</th>
<th>PHG (MCLG) [MRDLG]</th>
<th>UCD Groundwater</th>
<th>Sacramento River Water</th>
<th>Major Sources in Drinking Water (footnotes)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Aluminum</td>
<td>2017-2019</td>
<td>ppm</td>
<td>1</td>
<td>0.6</td>
<td>ND - 0.092</td>
<td>0.025</td>
<td>ND</td>
</tr>
<tr>
<td>Arsenic</td>
<td>2017-2019</td>
<td>ppb</td>
<td>10</td>
<td>0.004</td>
<td>ND - 5.5</td>
<td>2.35</td>
<td>ND</td>
</tr>
<tr>
<td>Barium</td>
<td>2017-2019</td>
<td>ppm</td>
<td>1</td>
<td>2</td>
<td>ND - 0.13</td>
<td>0.02</td>
<td>ND</td>
</tr>
<tr>
<td>Chromium (Total)</td>
<td>2017-2019</td>
<td>ppm</td>
<td>50</td>
<td>(100)</td>
<td>ND - 14</td>
<td>4.33</td>
<td>ND</td>
</tr>
<tr>
<td>Copper</td>
<td>2017-2019</td>
<td>ppm</td>
<td>(1.3)</td>
<td>0.3</td>
<td>ND</td>
<td>ND</td>
<td>0.0048</td>
</tr>
<tr>
<td>Fluoride</td>
<td>2017-2019</td>
<td>ppm</td>
<td>2.0</td>
<td>1</td>
<td>0.11 - 0.14</td>
<td>0.13</td>
<td>ND</td>
</tr>
<tr>
<td>Nitrate (as N)</td>
<td>2019</td>
<td>ppm</td>
<td>10</td>
<td>10</td>
<td>ND - 0.9</td>
<td>0.23</td>
<td>ND</td>
</tr>
<tr>
<td>Selenium</td>
<td>2017-2019</td>
<td>ppb</td>
<td>50</td>
<td>30</td>
<td>ND - 9.9</td>
<td>2.65</td>
<td>ND</td>
</tr>
</tbody>
</table>

#### Primary Drinking Water Standard Constituents - Inorganic Constituents

#### Primary Drinking Water Standard Constituents - Synthetic Organic Constituents

#### Primary Drinking Water Standard Constituents - Disinfection Byproducts

#### Primary Drinking Water Standard Constituents - Radioactive Constituents

#### Secondary Drinking Water Standard Constituents

**Note 1:** There are no PHGs or MCLGs for constituents with secondary drinking water standards because these are not health-based levels, but set on the basis of aesthetics.

### Table 2 - Detected Constituents Without A Drinking Water Standard

<table>
<thead>
<tr>
<th>Constituent</th>
<th>Year Sampled</th>
<th>Unit</th>
<th>UCD Groundwater</th>
<th>Sacramento River Water</th>
<th>Major Sources in Drinking Water (footnotes)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Alkalinity, Bicarbonate (as CaCO3)</td>
<td>2017-2019</td>
<td>ppm</td>
<td>200 - 220</td>
<td>210</td>
<td>66</td>
</tr>
<tr>
<td>Alkalinity, Carbonate (as CaCO3)</td>
<td>2017-2019</td>
<td>ppm</td>
<td>ND - 7.6</td>
<td>2</td>
<td>54</td>
</tr>
<tr>
<td>Calcium</td>
<td>2017-2019</td>
<td>ppm</td>
<td>14 - 27</td>
<td>17</td>
<td>12</td>
</tr>
<tr>
<td>Hardness, Total (as CaCO3)</td>
<td>2017-2019</td>
<td>ppm</td>
<td>80 - 190</td>
<td>120</td>
<td>55</td>
</tr>
<tr>
<td>Magnesium</td>
<td>2017-2019</td>
<td>ppm</td>
<td>11 - 29</td>
<td>18.67</td>
<td>6</td>
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<tr>
<td>pH</td>
<td>2017-2019</td>
<td>N/A</td>
<td>8.2 - 8.3</td>
<td>8.28</td>
<td>8</td>
</tr>
<tr>
<td>Sodium</td>
<td>2017-2019</td>
<td>ppm</td>
<td>73 - 88</td>
<td>71.83</td>
<td>17</td>
</tr>
</tbody>
</table>
# Disinfection By-Products

<table>
<thead>
<tr>
<th>Year Sampled</th>
<th>Unit</th>
<th>Regulatory Limits</th>
<th>Results Range</th>
<th>Major Sources in Drinking Water (footnotes)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>MCL</td>
<td>PHG or (MCLG)</td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>[4.0]</td>
<td>[4.0]</td>
<td>0.14 - 1.70 F</td>
</tr>
</tbody>
</table>

## Table 3 - Disinfection By-Products

1. **Total Trihalomethanes**: 2019 ppb 80 N/A ND - 36 F
2. **Total Haloacetic Acids**: 2019 ppb 60 N/A ND - 13 F
3. **Residual Chlorine**: 2019 ppm [4.0] [4.0] 0.14 - 1.70 F

## Table 4 - Total Coliform

<table>
<thead>
<tr>
<th>Year Sampled</th>
<th>Unit</th>
<th>Regulatory Limits</th>
<th>Samples Collected</th>
<th>Range Detected</th>
<th>Major Sources in Drinking Water (footnotes)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>MCL</td>
<td>MCLG</td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>5%</td>
<td>0%</td>
<td>631</td>
<td>U</td>
</tr>
</tbody>
</table>

## Table 4 - Lead and Copper

<table>
<thead>
<tr>
<th>Year Sampled</th>
<th>Unit</th>
<th>Regulatory Limits</th>
<th>No. of Samples Collected</th>
<th>90th Percentile Level Result</th>
<th>No. of Sites Exceeding Action Level</th>
<th>Major Sources in Drinking Water (footnotes)</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Action Level</td>
<td>PHG</td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td></td>
<td></td>
<td>15</td>
<td>0.2</td>
<td>61</td>
<td>ND</td>
<td>0 K</td>
</tr>
</tbody>
</table>

## Table 6 - Constituents Detected Under UCMR4

<table>
<thead>
<tr>
<th>Year Sampled</th>
<th>Unit</th>
<th>UCD Groundwater</th>
<th>Surface Water</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Range Detected</td>
<td>Average</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
<tr>
<td>Manganese²</td>
<td>2018</td>
<td>ppb</td>
<td>ND - 48</td>
</tr>
</tbody>
</table>

## Table 7 - Constituents Detected Under UCMR4

<table>
<thead>
<tr>
<th>Year Sampled</th>
<th>Unit</th>
<th>UCD Groundwater</th>
</tr>
</thead>
<tbody>
<tr>
<td></td>
<td></td>
<td>Results Range</td>
</tr>
<tr>
<td>Bromochloroacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Bromodichloroacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Chlorodibromoacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Dibromoacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Dichloroacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Monobromoacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Monochloroacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
<tr>
<td>Trichloroacetic acid</td>
<td>2018</td>
<td>ppb</td>
</tr>
</tbody>
</table>

## Note

² Although manganese is regulated under the California Code of Regulations, UCMR4 required testing of this constituent at a lower detection level.

## Unregulated Contaminant Monitoring Rule 4

As part of the Safe Drinking Water Act Amendments of 1996, the U.S. Environmental Protection Agency (EPA) is required to create a list every five years of up to 30 unregulated contaminants to be monitored in public water supplies. This list is derived from the Candidate Contaminant List (CCL) and represents compounds for which the EPA may consider as candidates for regulation. The University performed part of the sampling requirements for the Unregulated Contaminant Monitoring Rule 4 (UCMR4) in 2018 and completed sampling in 2019. The table below lists the unregulated constituents that were detected during UCMR4 sampling events.

## Important Health Information

Some people may be more vulnerable to contaminants in drinking water than the general population. Immuno-compromised persons such as persons with cancer undergoing chemotherapy, persons who have undergone organ transplants, people with HIV/AIDS or other immune system disorders, some elderly, and infants can be
particularly at risk from infections. These people should seek advice about drinking water from their health care providers. U.S. EPA/Centers for Disease Control (CDC) guidelines on appropriate means to lessen the risk of infection by Cryptosporidium and other microbial contaminants are available from the Safe Drinking Water Hotline (1-800-426-4791).

Cryptosporidium is a microbial pathogen found in surface water throughout the United States. Cryptosporidium was detected one time in the untreated surface water during 2019. However, the Regional Water Treatment Facility is designed to remove and/or deactivate these pathogens to the level that meets all drinking water standards. Note that current test methods approved by the USEPA do not distinguish between dead organisms and those capable of causing disease. Ingestion of Cryptosporidium may produce symptoms of nausea, abdominal cramps, diarrhea, and associated headaches. Cryptosporidium must be ingested to cause disease, and it may be spread through means other than drinking water.

Coliforms are bacteria that are naturally present in the environment and are used as an indicator that other, potentially-harmful, bacteria may be present. Coliforms were not found in any distribution system samples in 2019. While your drinking water meets the federal and state standard for arsenic, it does contain low levels of arsenic. The arsenic standard balances the current understanding of arsenic’s possible health effects against the costs of removing arsenic from drinking water. The U.S. Environmental Protection Agency continues to research the health effects of low levels of arsenic, which is a mineral known to cause cancer in humans at high concentrations and is linked to other health effects such as skin damage and circulatory problems.

If present, elevated levels of lead can cause serious health problems, especially for pregnant women and young children. Lead in drinking water is primarily from materials and components associated with service lines and home plumbing. UC Davis Utilities Division is responsible for providing high quality drinking water, but cannot control the variety of materials used in plumbing components. When your water has been sitting for several hours, you can minimize the potential for lead exposure by flushing your tap for 30 seconds to 2 minutes before using water for drinking or cooking. If you do so, you may wish to collect the flushed water and reuse it for another beneficial purpose, such as watering plants. If you are concerned about lead in your water, you may wish to have your water tested. Information on lead in drinking water, testing methods, and steps you can take to minimize exposure is available from the Safe Drinking Water Hotline or at [http://www.epa.gov/lead](http://www.epa.gov/lead).

**Water Conservation Tips for Consumers**

Did you know that the average U.S. household uses approximately 400 gallons of water per day or 100 gallons per person per day? Luckily, there are many low-cost and no-cost ways to conserve water. Small changes can make a big difference – try one today and soon it will become second nature.

- Take short showers – a 5 minutes shower uses 4 to 5 gallons of water compared to up to 50 gallons for a bath.
- Shut off water while brushing your teeth, washing your hair, and shaving and save up to 500 gallons a month.
- Use a water-efficient showerhead. They are inexpensive, easy to install, and can save you up to 750 gallons a month.
- Run your clothes washer and dishwasher only when they are full. You can save up to 1,000 gallons a month.
- Water plants only when necessary.
- Fix leaking toilets and faucets. Faucet washers are inexpensive and take only a few minutes to replace. To check your toilet for a leak, place a few drops of food coloring in the tank and wait. If it seeps into the toilet bowl without flushing, you have a leak. Fixing it or replacing it with a new, more efficient model can save up to 1,000 gallons a month.
- Adjust sprinklers so only your lawn is watered. Apply water only as fast as the soil can absorb it and during the cooler parts of the day to reduce evaporation.
- Teach your kids about water conservation to ensure a future generation that uses water wisely. Make it a family effort to reduce next month’s water bill!

Visit [www.epa.gov/watersense](http://www.epa.gov/watersense) for more information.

Source Water Protection
Protection of drinking water is everyone’s responsibility. You can help protect your community’s drinking water source in several ways:

- Eliminate excess use of lawn and garden fertilizers and pesticides – they contain hazardous chemicals that can reach your drinking water source.
- Pick up after your pets.
- If you have your own septic system, properly maintain your system to reduce leaching to water sources or consider connecting to a public water system.
- Dispose of chemicals properly; take used motor oil to a recycling center.
- Volunteer in your community. Find a watershed or wellhead protection organization in your community and volunteer to help. If there are no active groups, consider starting one. Use U.S. EPA’s Adopt Your Watershed to locate groups in your community, or visit the Watershed Information Network’s How to Start a Watershed Team.
- Organize a storm drain stenciling project with your local government or water supplier. Stencil a message next to the street drain reminding people “Dump No Waste – Drains to River” or “Protect Your Water”. Produce and distribute a flyer for households to remind residents that storm drains dump directly into your local water body.